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Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.009 Å R factor = 0.064 wR factor = 0.164 Data-to-parameter ratio = 23.0

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1-Allyl-4-aminopyridinium bromide

The asymmetric unit of the title structure, $C_8H_{11}N_2^+ \cdot Br^-$, contains two independent cations and two anions. In one cation, the dihedral angle between the mean planes of the pyridinium ring and the allyl group is 86.43 (6)°, while in the other cation this dihedral angle is 72.93 (7)°. In the crystal structure, anions and cations are linked *via* intermolecular $N-H\cdots$ Br hydrogen bonds, forming one-dimensional chains along [010].

Comment

In human peripheral mononuclear cells isolated from healthy volunteers, derivatives of pyridinium compounds have been identified. 4-Aminopyridine (4AP) has been found to be an efficient drug, affecting potassium permeability and capable of provoking membrane depolarization (Grissmer *et al.*, 1992) and is also able to induce an increase in intracellular calcium influx through modulation of the activity of purinergic cationic channels (Lajdova *et al.*, 2004). We report here the crystal structure of the title pyridinium derivative, (I).



The asymmetric unit of (I) consists of two independent pyridinium cations (labelled A and B) and two bromide anions (Fig. 1). The bond lengths and angles within the pyridinium ring are comparable with those in reported structures (Seethalakshmi, Kaliannan *et al.*, 2006; Seethalakshmi *et al.*, 2006*a*,*b*,*c*). The dihedral angle between the mean planes of the pyridinium ring and allyl group is 86.4 (6)° for cation A and 72.93 (7)° for cation B.

In the crystal structure, atom N2 of each cation acts as a donor for an intermolecular $N-H\cdots Br$ hydrogen bond with two different symmetry-related bromide anions (Table 1 and Fig. 2). These hydrogen bonds link cations *A* and *B* into a one-dimensional chain parallel to the *b* axis, which can be described by graph-set motif *C*(5).

Experimental

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A solution of 4-aminopyridine (1.175 g, 25 ml) and allyl bromide (1.51 g, 25 ml) in dry acetone (20 ml) was stirred for 44 h at room

Received 16 September 2006 Accepted 22 September 2006 temperature (303 K). The solid that separated was filtered, washed with dry acetone and dried in a vacuum to give the stable salt, (I), which was recrystallized from an aqueous ethanol (80% v/v) solution (yield 74%).

> V = 889.6 (5) Å³ Z = 4

 $D_x = 1.606 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation $\mu = 4.56 \text{ mm}^-$

Block, colourless

0.49 \times 0.32 \times 0.26 mm

9293 measured reflections

4585 independent reflections

 $w = 1/[\sigma^2(F_0^2) + (0.0581P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

-3

+ 8.4291P]

 $(\Delta/\sigma)_{\rm max} < 0.001$

 $\Delta \rho_{\text{max}} = 1.98 \text{ e Å}$ $\Delta \rho_{\rm min} = -1.25 \text{ e } \text{\AA}^{-3}$

3443 reflections with $I > 2\sigma(I)$

T = 293 (2) K

 $R_{\rm int} = 0.035$ $\theta_{\rm max} = 28.9^\circ$

Crystal data

 $C_8H_{11}N_2^+ \cdot Br^ M_r = 215.10$ Triclinic, P1 a = 6.9085 (19) Åb = 9.179(3) Å c = 14.186 (4) Å $\alpha = 90.096 (4)^{\circ}$ $\beta = 97.446 \ (4)^{\circ}$ $\gamma = 94.164 \ (4)^{\circ}$

Data collection

Bruker APEX-2 CCD area-detector diffractometer and a scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\min} = 0.171, \ T_{\max} = 0.307$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.064$ $wR(F^2) = 0.164$ S = 1.064585 reflections 199 parameters H-atom parameters constrained

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$	1
$N2A - H2A1 \cdots Br1^{i}$	0.86	2.57	3.411 (6)	165	
$N2A - H2A2 \cdot \cdot \cdot Br2^{ii}$	0.86	2.58	3.399 (6)	159	
$N2B - H2B1 \cdots Br2^{iii}$	0.86	2.54	3.386 (5)	167	
$N2B - H2B2 \cdot \cdot \cdot Br1^{iii}$	0.86	2.51	3.354 (6)	169	
Symmetry codes: (i)	-x+1, -y+2, -z; (ii) $-x+1, -y+1, -z;$		+1, -z; (iii)	

-x + 1, -y + 1, -z + 1.

H atoms were positioned geometrically (C-H = 0.93 - 0.99 Å and N-H = 0.86 Å) and refined as riding, with $U_{iso}(H) = 1.2U_{eq}(C,N)$. The largest peak in the final difference Fourier is 0.92 Å from atom Br1 and the deepest hole is 0.79 Å from atom H4A.

Data collection: APEX2 (Bruker, 2006); cell refinement: APEX2; data reduction: APEX2; program(s) used to solve structure: SHELXL97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PARST (Nardelli, 1995).

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References

Bruker (2006). APEX2. Version 2.0-2. Bruker AXS Inc., Madison, Wisconsin, USA.

Grissmer, S., Lewis, R. S. & Cahalan, M. D. (1992). J. Gen. Physiol. 99, 63-84.



Figure 1

View of the aymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are represented by circles of arbitrary radii.



Figure 2

The crystal structure of (I), viewed along the a axis. Intermolecular N-H...Br hydrogen bonds are indicated by dashed lines.

Lajdova, I., Chorvat, D. Jr, Spustova, V. & Chorvatova, A. (2004). Can. J. Physiol. Pharmacol. 82, 50-56.

Nardelli, M. (1995). J. Appl. Cryst. 28, 659.

- Seethalakshmi, T., Kaliannan, P., Venkatesan, P., Fronczek, F. R. & Thamotharan, S. (2006). Acta Cryst. E62, o2353-o2355.
- Seethalakshmi, T., Venkatesan, P., Fronczek, F. R., Kaliannan, P. & Thamotharan, S. (2006a). Acta Cryst. E62, o2560-o2562.
- Seethalakshmi, T., Venkatesan, P., Fronczek, F. R., Kaliannan, P. & Thamotharan, S. (2006b). Acta Cryst. E62, 03389-03390.
- Seethalakshmi, T., Venkatesan, P., Butcher, R. J., Nallu, M. & Kaliannan, P (2006c). Acta Cryst. E62, 04010-04011.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.